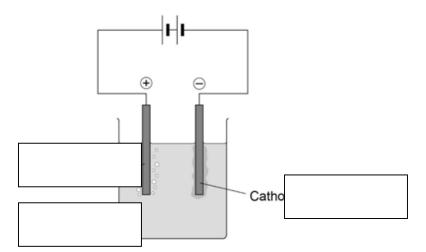
## **Chemistry: Required Practical 2**





What is this required practical about?

**€** Label the diagram

Think about how you carried out this practical.

Write a logical and comprehensive method, using numbered bullet points.

Is your method good?

Read it. If someone else could get results from it then yes!

Why is it important that the "rods" do not touch each other?	
<b>₡</b> What are the "rods" made of?	
<b>₡</b> What might you see happening at the cathode?	
<b>ば</b> What might you see happening at the anode?	
List some other observations you could make that would indicate a reaction	occurring.
Explain how electrolysis can split ionic compounds, like lead bromide.	
■ During electrolysis of lead bromide, outline what happens at the anode and	
# HT ONLY: write the half equations for what is happening at the anode and t	ne cathode.